

# Total Synthesis of Tetrahydroquinoline Alkaloid (+)-Angustureine<sup>†</sup>

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(+)-(*S*)-Angustureine was synthesized using a CuI-catalyzed coupling reaction of iodobenzene with an enantiopure  $\beta$ -amino ester as the key step. The overall yield is 36% in 7 linear steps.

**Keywords**    alkaloid, total synthesis, coupling

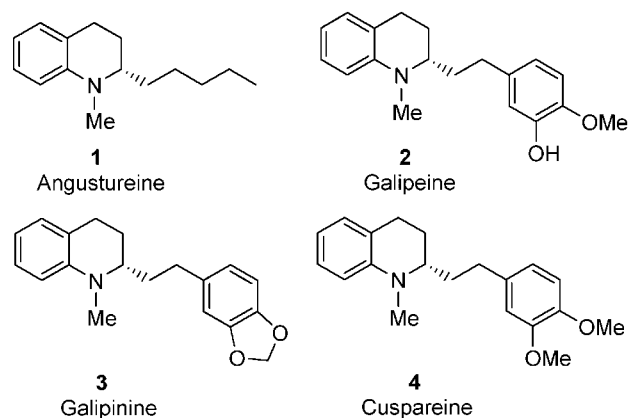
## Introduction

Angustureine (**1**), galipeine (**2**), galipinine (**3**) and cuspareine (**4**) are four 2-substituted 1-methyltetrahydroquinoline alkaloids isolated from the bark of *Galipea officinalis* Hancock by professor Ingrid Jacquemond-Collet *et al.* in 1999.<sup>1</sup> *Galipea officinalis* Hancock is a shrub indigenous to the mountains of venezuela that has marked influence on the spinal motor nerves, thus, serves a cure for paralytic affections.<sup>1</sup> Further biological testing indicated that these four compounds possessed promising in vitro antimalarial activity with IC<sub>50</sub> values from 0.09 to 38 mg/mL for the chloroquine-resistant strains.<sup>2</sup> Therefore, these alkaloids can serve as new leads for development of new antimalarial drugs. Very recently, Zhou and his co-workers have reported the first synthesis and stereochemical assignments of (–)-angustureine (**1**), (–)-galipinine (**3**) and (–)-cuspareine (**4**) (Figure 1) by using their newly developed enantioselective hydrogenation of substituted quinolines.<sup>3</sup> Herein we wish to disclose a facile route to these compounds exemplified by total synthesis of unnatural (+)-angustureine (**1**).

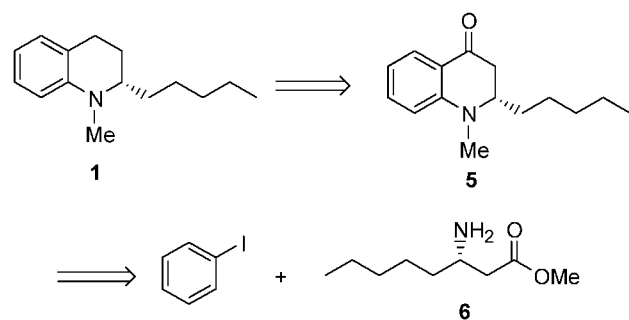
## Results and discussion

As demonstrated in Figure 2, we planned to synthesize the target molecule through (*S*)-1-methyl-2-pentyl-2,3-dihydro-1*H*-quinolin-4-one (**5**) by Pd/C-catalyzed reduction of the ketone moiety. Based on our previous studies,<sup>4</sup> **5** could be assembled from a sequential CuI-catalyzed coupling product of iodobenzene with an enantiopure  $\beta$ -amino ester **6** and intramolecular acylation process.

As outlined in Scheme 1, our synthesis started from a Micheal addition reaction of lithium (*S*)-*N*-benzyl- $\alpha$ -



**Figure 1** Structures of angustureine, galipeine, galipinine and cuspareine.



**Figure 2** Retrosynthetic analysis of (+)-angustureine (**1**).

methylbenzylamide to 2-octenoic acid methyl ester **7** to afford  $\beta$ -amino ester **8** following Davies's procedure.<sup>5</sup> After Pd/C-catalyzed hydrogenolysis of **8** in acetic acid, the resulting free amine **6** was subjected to a CuI-catalyzed coupling reaction with iodobenzene to provide *N*-phenyl  $\beta$ -amino acid **9** in 86% yield. This intermediate was also prepared via a *L*-proline-promoted CuI-

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catalyzed coupling<sup>6</sup> of **6** and iodobenzene at 60 °C followed by hydrolysis of the ester moiety with aqueous K<sub>2</sub>CO<sub>3</sub>. The second method was obviously more suitable for some substrates with less stable functional groups in later SAR studies.

Introduction of a methyl group to N-1 was successfully accomplished with a S<sub>N</sub>2 reaction of iodomethane with **9** in the presence of Ag<sub>2</sub>O. At this time the carboxylate moiety was also esterified to give **10**. Hydrolysis of **10** with aqueous NaOH in methanol produced acid **11**. Next, treatment of **11** with thionyl chloride to provide the corresponding acyl chloride, which was exposed to AlCl<sub>3</sub> in methylene chloride to furnish the ketone **5** in 77% yield. Finally, hydrogenation of **5** worked well under the catalysis of Pd/C to afford (S)-angustureine (**1**) (Scheme 2) ( $[\alpha]_{\text{D}}^{20} +8.0$  (c 1.0, CHCl<sub>3</sub>); lit.<sup>1</sup>:  $[\alpha]_{\text{D}}^{20} -7.16$  for natural angustureine; lit.<sup>3</sup>:  $[\alpha]_{\text{D}}^{20} -6.7$  (c 1.0, CHCl<sub>3</sub>) for (−)-angustureine in 94%). The overall yield was 36% over 7 steps.

As a summary, we have developed a facile route to enantiopure 2-substituted 1-methyltetrahydroquinoline alkaloids using the CuI-catalyzed coupling reaction of aryl iodides with β-amino esters as the key steps. Since a variety of these two intermediates are conveniently available, the present protocol would allow to assemble

a wide range of 2-substituted 1-methyltetrahydroquinoline alkaloids for biological evaluation. Further studies toward this direction are underway.

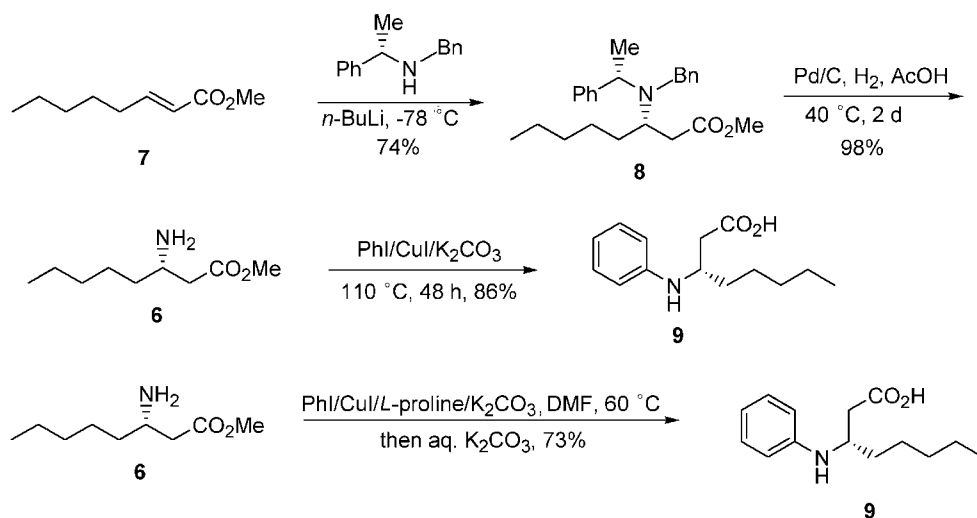
## Experimental

IR spectra were measured on a Shimadzu 440 spectrometer. <sup>1</sup>H NMR spectra were recorded with TMS as an internal standard on a Bruker AM-300 or Varian EM-360. MS spectra were determined on an HP-5989A spectrometer. HRMS (MALDI) spectra were obtained on a IonSpec FTMS. Optical rotations were obtained on a Perkin-Elmer 241 Autopol polarimeter. All reactions were run in flame-dried glassware under an argon atmosphere unless stated otherwise.

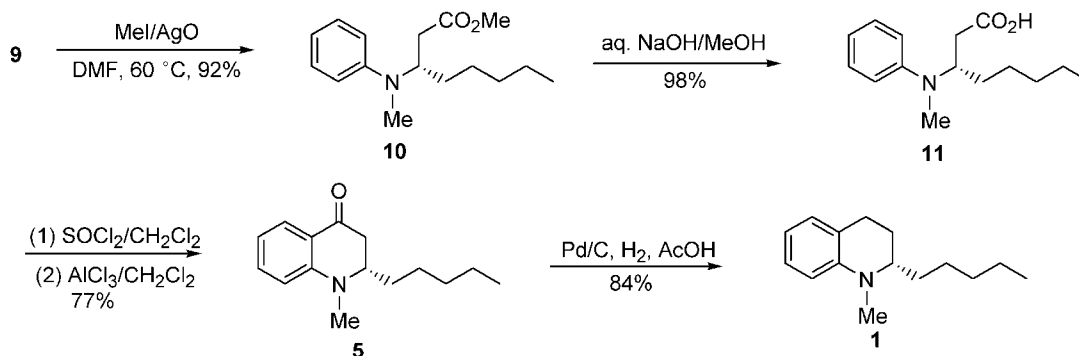
**(S)-3-[Benzyl-(1-phenylethyl)amino]octanoic acid methyl ester (8)**<sup>7</sup>:  $[\alpha]_{\text{D}}^{20} -9.9$  (c 1.2, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz) δ: 7.44–7.21 (m, 10H), 3.85 (d, *J*=13.5 Hz, 1H), 3.83 (q, *J*=6.3 Hz, 1H), 3.72 (d, *J*=13.5 Hz, 1H), 3.53 (s, 3H), 3.27–3.28 (m, 1H), 2.03 (d, *J*=6.0 Hz, 2H), 1.55–1.57 (m, 2H), 1.35–1.11 (m, 9H), 0.88 (t, *J*=6.6 Hz, 3H); IR (KBr) ν: 1737, 1150 cm<sup>-1</sup>; ESI-MS *m/z* 368 (M+H)<sup>+</sup>.

**(S)-3-Aminooctanoic acid methyl ester (6)**<sup>7</sup>:  $[\alpha]_{\text{D}}^{20} +11.5$  (c 1.2, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz) δ:

Scheme 1



Scheme 2



3.70 (s, 3H), 3.21—3.16 (m, 1H), 2.48 (dd,  $J=15.6$ , 3.9 Hz, 1H), 2.26 (dd,  $J=15.6$ , 5.9 Hz, 1H), 1.40—1.28 (m, 8H), 0.88 (t,  $J=6.6$  Hz, 3H); IR (KBr)  $\nu$ : 3302, 1737, 1639  $\text{cm}^{-1}$ ; ESI-MS  $m/z$  174 ( $M+H$ )<sup>+</sup>.

**(S)-3-(N-Phenyl)aminoctanoic acid (9)**: To a mixture of **6** (1.37 g, 8.6 mmol), potassium carbonate (3.57 g, 25.8 mmol), CuI (164 mg, 0.86 mmol), 1 mL of H<sub>2</sub>O and 20 mL of DMF was added iodobenzene (1.93 mL, 17.26 mmol) dropwise. The resulting suspension was heated at 110 °C for 48 h. The cooled reaction mixture was acidified with 1 mol/L HCl to pH=2, and then extracted with ethyl acetate. The combined organic layers were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. The residual oil was chromatographed eluting with 1 : 3 (ethyl acetate/petroleum ether) to afford **9** (1.56 g, 86%).  $[\alpha]_D^{20} +1.8$  ( $c$  1.0, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$ : 7.20 (t,  $J=7.5$  Hz, 2H), 6.76 (t,  $J=7.5$  Hz, 1H), 6.67 (d,  $J=7.5$  Hz, 2H), 3.78 (m, 1H), 2.58 (dd,  $J=6.0$ , 1.8 Hz, 2H), 1.59 (q,  $J=6.6$  Hz, 2H), 1.45—1.35 (m, 2H), 1.35—1.24 (m, 4H), 0.87 (t,  $J=6.6$  Hz, 3H); IR (KBr)  $\nu$ : 3377, 3054, 1709, 1603  $\text{cm}^{-1}$ ; EI-MS  $m/z$  235 ( $M^+$ ). HRMS (MALDI) calcd for C<sub>14</sub>H<sub>22</sub>NO<sub>2</sub> ( $M+H$ )<sup>+</sup> 236.1645, found 236.1656.

**(S)-3-(N-Methyl-N'-phenyl)aminoctanoic acid methyl ester (10)**: A mixture of **9** (2.3 g, 7.3 mmol), Ag<sub>2</sub>O (774 mg, 29.2 mmol) and iodomethane (3.6 mL, 58.4 mmol) in 15 mL of DMF was stirred at room temperature for 3 h, and then allowed to warm to 60 °C overnight. After the starting material disappeared as monitored by TLC, the reaction mixture was filtered and the filtrate was concentrated *in vacuo*. The crude product was purified by flash chromatography on silica gel eluting with 1 : 50 (ethyl acetate/petroleum ether) to afford **10** (2.3 g, 92%).  $[\alpha]_D^{20} -18.6$  ( $c$  1.2, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$ : 7.22 (t,  $J=7.5$  Hz, 2H), 6.84 (d,  $J=7.6$  Hz, 2H), 6.70 (t,  $J=7.5$  Hz, 1H), 4.31—4.33 (m, 1H), 3.55 (s, 3H), 2.71 (s, 3H), 2.60 (dd,  $J=14.1$ , 8.1 Hz, 1H), 2.44 (dd,  $J=14.1$ , 8.1 Hz, 1H), 1.71—1.55 (m, 1H), 1.55—1.41 (m, 1H), 1.35—1.15 (m, 1H), 0.85 (t,  $J=6.9$  Hz, 3H); IR (KBr)  $\nu$ : 1740, 1600  $\text{cm}^{-1}$ ; EI-MS  $m/z$  263 ( $M^+$ ), 192, 160, 132. HRMS (MALDI) calcd for C<sub>16</sub>H<sub>26</sub>NO<sub>2</sub> ( $M+H$ )<sup>+</sup> 264.1958, found 264.1964.

**(S)-3-(N-Methyl-N'-phenyl)aminoctanoic acid (11)**: A mixture of **10** (472 mg, 1.8 mmol) and NaOH (360 mg, 9 mmol) in the mixed solvent of methanol (5 mL) and water (5 mL) was heated at 60 °C for 2 h. The cooled solution was acidified with 1 mol/L HCl to pH=2 and extracted thoroughly with ethyl acetate. The extract was washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. The residual oil was chromatographed eluting with 1 : 4 (ethyl acetate/petroleum ether) to afford **11** (417 mg, 98%).  $[\alpha]_D^{20} -8.2$  ( $c$  1.0, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$ : 7.26 (t,  $J=7.5$  Hz, 2H), 6.98 (d,  $J=7.5$  Hz, 2H), 6.87 (t,  $J=7.5$  Hz, 1H), 4.11—4.13 (m, 1H), 2.75 (s, 3H), 2.70—2.45 (m, 2H), 2.16—1.54 (m, 1H), 1.50—1.40 (m, 1H), 1.31—1.22 (m, 6H), 0.87 (t,  $J=6.4$  Hz, 3H); IR (KBr)  $\nu$ : 1708, 1599  $\text{cm}^{-1}$ ;

EI-MS  $m/z$  249 ( $M^+$ ). HRMS (MALDI) calcd for C<sub>15</sub>H<sub>24</sub>NO<sub>2</sub> ( $M+H$ )<sup>+</sup> 250.1802, found 250.1806.

**1-Methyl-2-pentyl-2,3-dihydro-1H-quinolin-4-one (5)**: To a solution of **10** (379 mg, 1.6 mmol) in 10 mL of anhydrous CH<sub>2</sub>Cl<sub>2</sub> at 0 °C was added thionyl chloride (0.7 mL, 8 mmol) and then allowed to stir for 2 h at room temperature. The solvent was removed *in vacuo* to afford a brown oil residue, which was dissolved in 10 mL of anhydrous CH<sub>2</sub>Cl<sub>2</sub> followed by the addition of AlCl<sub>3</sub> (1.05 g, 8 mmol). The mixture was stirred at room temperature for 24 h before the reaction was quenched with 10 mL of water. The organic layer was separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. The residual oil was chromatographed eluting with 1 : 50 (ethyl acetate/petroleum ether) to afford **5** (280 mg, 77%).  $[\alpha]_D^{20} -199.8$  ( $c$  1.0, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$ : 7.85 (d,  $J=7.5$  Hz, 1H), 7.39 (t,  $J=7.5$  Hz, 1H), 6.67 (t,  $J=7.5$  Hz, 1H), 6.60 (d,  $J=7.5$  Hz, 1H), 3.49—3.51 (m, 1H), 3.03 (s, 1H), 2.97 (dd,  $J=15.9$ , 6.2 Hz, 1H), 2.65 (dd,  $J=15.9$ , 2.4 Hz, 1H), 1.58—1.52 (m, 2H), 1.35—1.39 (m, 6H), 0.86 (t,  $J=6.8$  Hz, 3H); IR (KBr)  $\nu$ : 1675, 1604  $\text{cm}^{-1}$ ; EI-MS  $m/z$  231 ( $M^+$ ). HRMS (MALDI) calcd for C<sub>15</sub>H<sub>22</sub>NO ( $M+H$ )<sup>+</sup> 232.1696, found 232.1707.

**(S)-Angustureine (1)**: A mixture of **5** (93 mg, 0.4 mmol) and 10% Pd/C (30 mg) in 5 mL of methanol and 1 mL of acetic acid was stirred at room temperature overnight under 101 kPa H<sub>2</sub>. After removing Pd/C by filtration, the filtrate was concentrated and then purified by flash chromatography eluting with petroleum ether to give **1** (74 mg, 84%).  $[\alpha]_D^{20} +8.0$  ( $c$  1.0, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$ : 7.08 (t,  $J=7.5$  Hz, 1H), 6.97 (d,  $J=7.5$  Hz, 1H), 6.57 (t,  $J=7.5$  Hz, 1H), 6.51 (t,  $J=7.5$  Hz, 1H), 3.22—3.24 (m, 1H), 2.92 (s, 3H), 2.80—2.62 (m, 2H), 1.91—1.84 (m, 2H), 1.42—1.22 (m, 6H), 0.89 (t,  $J=6.9$  Hz, 3H); IR (KBr)  $\nu$ : 2955, 1604  $\text{cm}^{-1}$ ; EI-MS  $m/z$  217 ( $M^+$ ). HRMS (MALDI) calcd for C<sub>15</sub>H<sub>24</sub>N ( $M+H$ )<sup>+</sup> 218.1903, found 218.1917.

## References

- Jacquemond-Collet, I.; Hannedouche, S.; Fabre, N.; Fourasté, I.; Moulis, C. *Phytochemistry* **1999**, *51*, 1167.
- Jacquemond-Collet, I.; Benoit-Vical, F.; Valentin, M. A.; Stanislas, E.; Mallie, M.; Fouraste, I. *Planta Med.* **2002**, *68*, 68.
- Wang, W.-B.; Lu, S.-M.; Yang, P.-Y.; Han, X.-W.; Zhou, Y.-G. *J. Am. Chem. Soc.* **2003**, *125*, 10536.
- (a) Ma, D.; Xia, C. *Org. Lett.* **2001**, *3*, 2583.  
(b) Ma, D.; Jiang, J.; *Tetrahedron: Asymmetry* **1998**, *9*, 1137.
- Davies, S. G.; Ichihara, O. *Tetrahedron: Asymmetry* **1991**, *2*, 183.
- Ma, D.; Cai, Q.; Zhang, H. *Org. Lett.* **2003**, *5*, 2453.
- Ma, D.; Pu, X.; Wang, J. *Tetrahedron: Asymmetry* **2002**, *13*, 2257.